

Chemistry 1110 R11
Fall 2001
Test #2

Thursday, October 25, 2001

Time: 1 hour 50 minutes

Name: _____

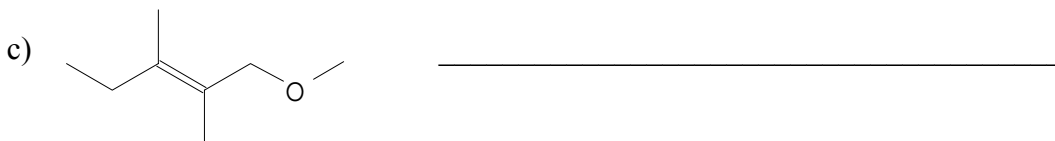
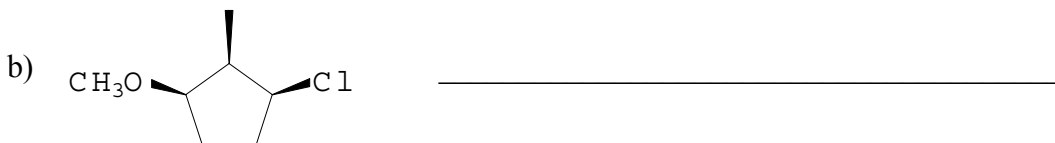
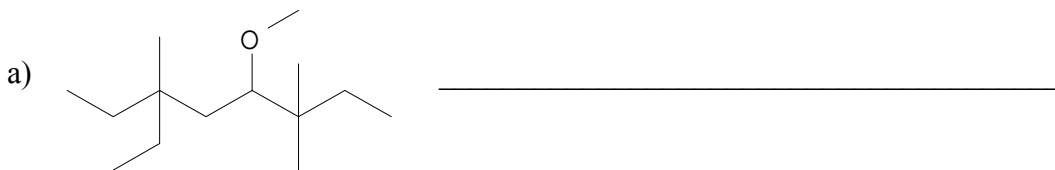
Student number: _____

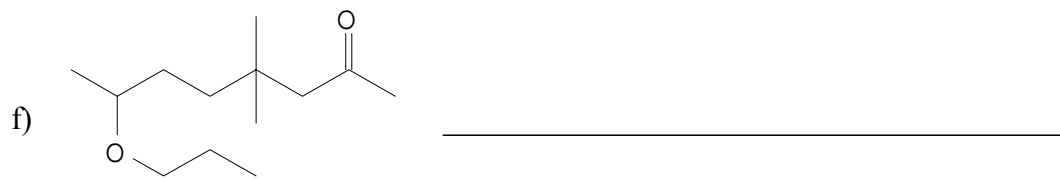
*This exam consists of **seven** pages of questions and a page with names and priorities of the functional groups. Please ensure that you have a complete paper and, if you do not, obtain one from me **immediately**. Good luck!*

Neatness counts! Remember that you are communicating information when you draw structures, and if I can't read what you write because it's messy, I can't tell if you're right or not. In cases of ambiguity (because of messy work), I have no choice but to assume you're incorrect.

Note: Only chemicals written above reaction arrows may be presumed to be present in excess.

1) [8 marks] Name the following compounds:





2) **[8 marks]** Draw the compounds that correspond to the following names:

a) 4,4-dimethylhexyl 3-chloropropanoate

b) N-(1-bromoethyl)-2-chlorohexanamide

c) N,N-dicyclopropylbutanamine

d) o-dichlorobenzene

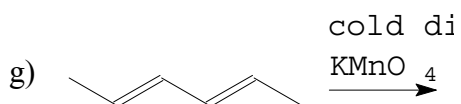
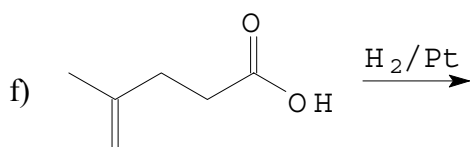
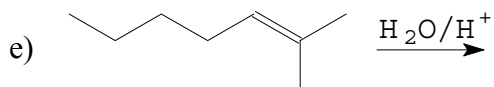
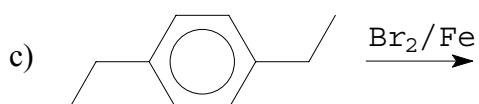
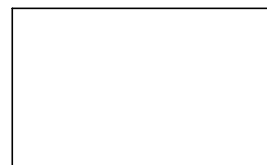
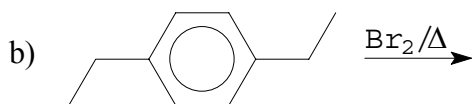
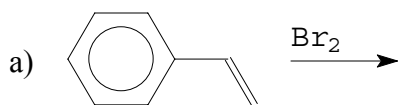
e) 4-nitrotoluene

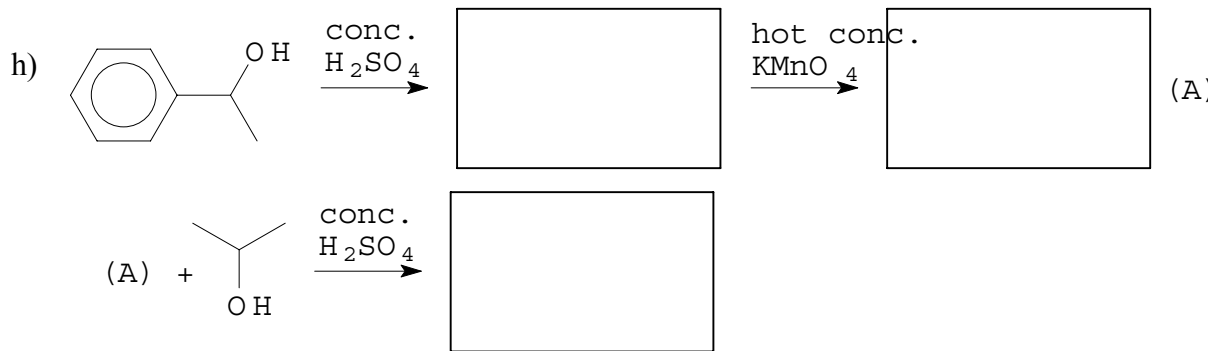
f) 4-chloro-2-methoxybenzoic acid

g) 2,4,6-trinitrophenol

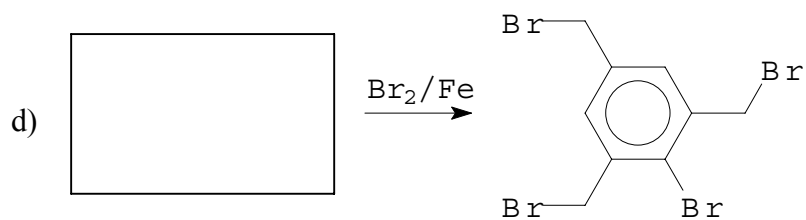
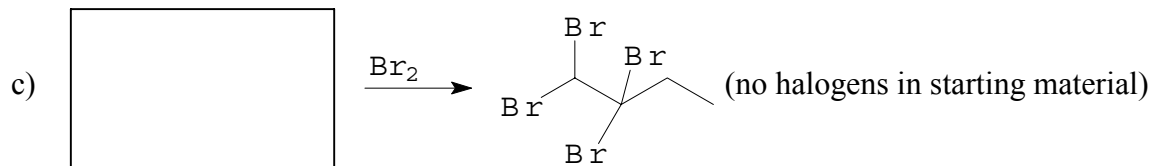
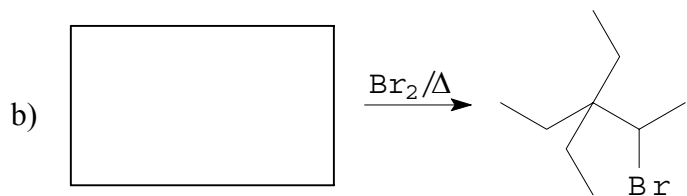
h) o-fluorophenyl ethanoate

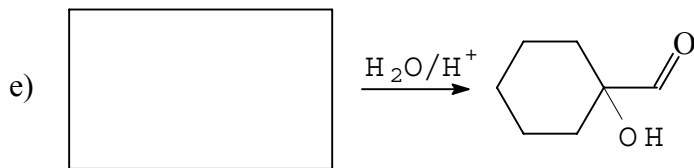
3) [10 marks total] Complete the following reactions. Give the structure of the organic product or products only.





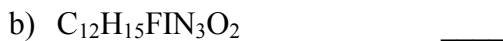
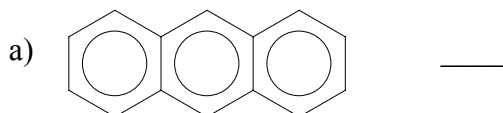
4) [8 marks] Provide the single organic reagent necessary to obtain the indicated organic products.





f) **[3 marks]** Starting from benzene, and using any other organic and inorganic reagents required, show how you would make TNT.

5) **[2 marks]** How many units of unsaturation are present in:



6) **[9 marks total]** Using the formula C_4H_8O and drawing only *saturated compounds*, answer the following questions:

a) **[3 marks]** Draw all possible functional, skeletal and positional isomers.

b) **[2 marks]** Pick a compound from (a) that is capable of geometric isomerism. Draw and label the cis and trans forms.

c) **[2 marks]** Pick a compound from (a) that is capable of optical isomerism. Put an asterisk beside the chiral carbon atom or atoms.

d) **[2 marks]** Draw a pair of functional isomers.

7) **[9 marks total]** Using the formula C_4H_9N , answer the following questions:

a) **[1 mark]** Draw a compound that will not react with H_2 .

b) **[2 marks]** Draw a pair of saturated positional isomers.

c) **[2 marks]** Draw a saturated compound that cannot have positional isomers.

d) **[2 marks]** Draw one molecule that will react with $KMnO_4$ and one that will not. Circle the molecule that will react with $KMnO_4$.

Functional Group Priorities and Prefix/Suffix Labels
(in order of functional group priority)

Compound Type	Prefix	Suffix
Carboxylic Acid	N/A	oic acid
Ester	N/A	oate
Amide	N/A	amide
Nitrile	cyano	nitrile
Aldehyde	oxo	al
Ketone	oxo	one
Alcohol	hydroxy	ol
Amine	amino	amine
Alkene	enyl	ene
Alkyne	ynyl	yne
Alkyl halide	halo	N/A
Ether	oxy	ether